

Rational Design and Evaluation of New Lead Compound Structures for Selective β ARK1 Inhibitors

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β -Adrenergic receptor kinase 1 (β ARK1) and cyclic adenosine 5'-monophosphate-dependent protein kinase A (PKA) have structurally similar adenine-binding pockets but have different physiologic functions. To obtain specific β ARK1 inhibitors, a two step rational drug design process was used. First, a search was conducted on three-dimensional models of commercially available compounds to find compounds that fit the adenine-binding pocket of β ARK1. Second, a comparative docking study that focused on the differences between the adenine-binding pockets of the two enzymes was used to evaluate the binding specificity of each compound that inhibited β ARK1 activity. The results of these analyses yielded three β ARK1-selective inhibitor leads from 11 candidates, a hit rate for selectivity of 27%. Although the IC_{50} values of these compounds for β ARK1 ranged from only 1.3×10^{-4} M to 5.6×10^{-4} M, the compounds did not inhibit PKA at concentrations up to 1.0×10^{-3} M. Thus, the present study shows the usefulness of a rational drug design strategy in finding specific kinase inhibitors for proteins with similar drug target binding sites.

Introduction

A rational drug design process employs many techniques. These techniques, which include three-dimensional (3D) database searches and pharmacophore searches, have been successfully applied to select ligand candidates from chemical compound databases.^{1–11} The present study uses a program named ARCHER, which was designed based on such algorithms,^{12–15} to search for selective inhibitor leads of β -adrenergic receptor kinase 1 (β ARK1). In addition to these techniques, Perola et al. objectively demonstrated the value or validity of their computational screening using a chemical database with a negative control study.¹ The present study also follows the negative control screening methodology of Perola et al. in order to assess the validity of the screening and design methods for finding selective inhibitor leads of β ARK1.

β ARK1 (EC 2.7.7.126) is a member of the G protein-coupled receptor kinase family.¹⁶ Proteins that belong to this family affect a wide range of physiologic functions and participate in the pathogenesis of many chronic systemic disorders.¹⁷ Indeed, by phosphorylating serine and threonine residues in the intracellular carboxyl terminal of β_2 -adrenergic receptors bound to receptor agonists, β ARK1 desensitizes β_2 -adrenergic receptors,¹⁸ which is thought to contribute to the development of diseases such as congestive heart failure.^{19–21} Thus, β ARK1 inhibitors might prove to be powerful new drugs. However, any such compounds might inhibit other kinases, since many of these kinases are thought to have a structural similarity to β ARK1. Therefore, the search

for specific β ARK1 inhibitors would benefit greatly from a rational, rather than random, screening and design process.

To conduct a computational drug screen properly, information on the tertiary structure and ligand recognition mechanism of the target molecule, in this case β ARK1, is essential. Because this information is unknown, a previously proposed 3D model of the β ARK1 catalytic domain was used.²² This structure is based on a crystalline structure of cyclic adenosine 5'-monophosphate (AMP)-dependent protein kinase A (PKA; EC 2.7.1.37)–PKI5-24 complex, registered in the Protein Data Bank as 2CPK.²³ PKA is also a serine–threonine kinase. β ARK1 and PKA share a 34.4% amino acid sequence identity in the catalytic domain. Additionally, PKA causes the desensitization of β_2 -adrenergic receptors, as β ARK1 does.²⁴ Furthermore, the amino acid residues at the adenosine 5'-triphosphate (ATP) binding site are highly conserved between β ARK1 and PKA.²² Consequently, the structure of PKA was used as a guide to construct a computer-generated 3D model of β ARK1.

PKA specifically recognizes the $p-11$, $p-6$, $p-3$, $p-2$, and $p+1$ positions of the peptide inhibitor.²⁵ These positions are named in reference to Ala21, the phosphorylation site of peptide substrates designated p . Results from a previous study suggest that the PKA amino acid residues involved in the specific recognition of PKI5-24 at $s-11$, $s-6$, $s-2$, and $s+1$ sites are not conserved in β ARK1 if the sites of PKA interacting with $p-n$ of PKI5-24 are designated as $s-n$.²² Consequently, the specific ligand recognition sites of β ARK1 were proposed to be $s-11$, $s-6$, $s-2$, and $s+1$.

In the present study, the strategy used to select potent and selective kinase inhibitors was to design compounds that interact with both the ligand recognition site and the adenine-binding pocket, a target site with high

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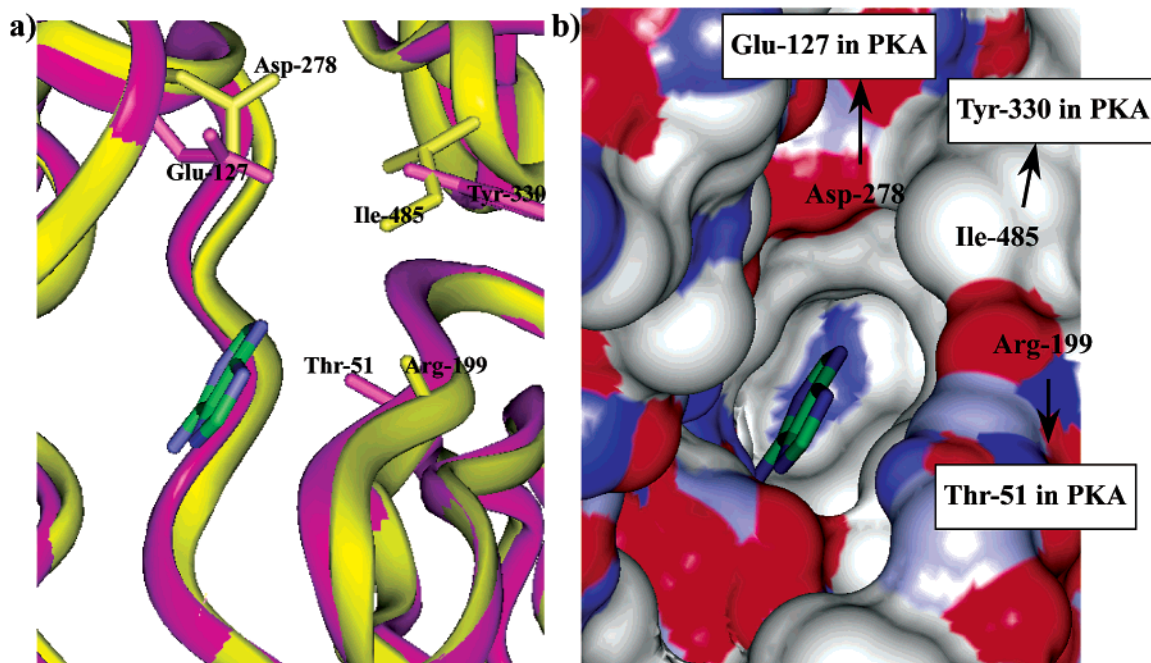


Figure 1. Ligand binding site with an adenine placed in reference to the PKA/PKI5-24/MgATP crystal structure.⁵⁸ (a) Comparison between β ARK1 model structure and PKA crystal structure. Yellow, β ARK1; pink, PKA. (b) Connolly surface drawings of β ARK1 model structure.

Table 1. Distance from the Adenine Binding Site to Each Ligand Recognition Site in the β ARK1 Model

ligand recognition site	representative atom	distance from Asp-272 ^a (Å)
<i>s</i> -11	Ser-389 O _γ	27.3
<i>s</i> -6	Gly-355 C _α	25.6
<i>s</i> -3	Asp-278 C _γ	11.2
<i>s</i> -2	Lys-319 N _ζ	17.3
<i>s</i> +1	His-354 N _{ε2}	27.6

^a Distances relative to the oxygen atom present in the Asp-272 main chain.

affinity for adenine in other kinases.^{26,27} To use this strategy with a molecule as small as the candidate drug, the two sites cannot be too far apart from each other. However, the specific ligand recognition sites in β ARK1, *s*-11, *s*-6, *s*-2, and *s*+1 are not suitable for designing selective inhibitor leads since, as listed in Table 1, they are located too far from the adenine-binding pocket of the ATP binding site. Therefore, another candidate site was needed.

In contrast with the *s*-11, *s*-6, *s*-2, and *s*+1 sites, the *s*-3 site is located next to the adenine-binding pocket in the previously proposed β ARK1 model.²² In this model, the *s*-3 site of β ARK1 is surrounded by the main chain carbonyl group of Arg-199 and side chains of Asp-278 and Ile-485 (Figure 1).²⁸ The corresponding site in PKA is surrounded by the main chain carbonyl group of Thr-51 and side chains of Glu-127 and Tyr-330. The steric capacity and shape of β ARK1 and PKA differ at this site in the following ways. First, Asp-278 in β ARK1 and Glu-127 in PKA differ from each other in their side chain lengths. Second, there is a steric difference in van der Waals volume caused by the difference in side chains between Ile-485 in β ARK1 and the corresponding Tyr-330 in PKA. Third, Thr-51 of PKA is slightly closer to Glu-127 than Arg-199 of β ARK1 is to Asp-278. Consequently, a small molecule that can

recognize these steric differences between β ARK1 and PKA holds the potential to be a specific β ARK1 inhibitor.

However, because the virtual screening method used in the present study does not directly compare the two enzymes, lead compound candidates selected by the virtual screening alone would not necessarily be specific to β ARK1. To estimate their specificity to β ARK1, comparative docking studies were used to find lead compound candidates with greater specificity for β ARK1. This comparative docking study used superimposition of PKA structure on the putative β ARK1-candidate complexes and qualitative estimation of repulsion between PKA and these candidates.

These facts and methods were used to form a rational drug design strategy to find specific β ARK1 inhibitors and to confirm their activities. This process consisted of four steps: (i) potent β ARK1 inhibitor lead compounds were found by pharmacophore-based virtual screening of a chemical structure database, (ii) the β ARK1 inhibitory activity of these compounds was tested by an *in vitro* assay, (iii) the specificity of the candidate β ARK1 inhibitors was tested by further docking studies with both β ARK1 and PKA structures, and (iv) the PKA inhibitory activity of these compounds was tested by an *in vitro* assay.

Results

Pharmacophore Generation. In the process of pharmacophore probe generation, a hydrogen bond donor (HBD), a hydrogen bond acceptor (HBA), and an aromatic ring centroid (ARC) were defined after referring to the interaction between other kinases and their ligands. The final, simplified β ARK1 pharmacophore model is shown in Figure 2. These pharmacophore probes were first used alone in the pharmacophore screening step and next used in combination with the

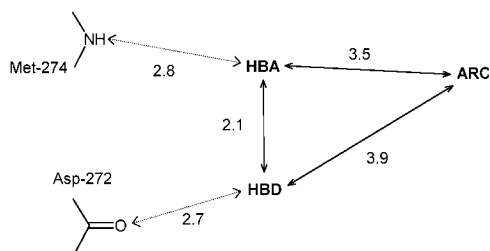


Figure 2. Schematic diagram outlining the ideal properties of a β ARK1 pharmacophore. Numbers represent distances in Ångstroms.

Table 2. χ Angle (Reflecting Torsion around C_{α} - C_{β} Bond) of Four Amino Acid Residues in the Two Conformers of β ARK1 and Their Corresponding Residues in the PKA Crystal Structure^a

	Ile-197	Leu-271	Ser-334	Phe-483
β ARK1 conformer 1	-70.2 (gauche+)	40.1 (gauche-)	-164.2 (trans)	-69.5 (gauche+)
β ARK1 conformer 2	227.1 (trans)	-58.2 (gauche+)	-68.2 (gauche+)	-61.3 (gauche+)
PKA crystal structure	Leu-49 -58.4 (gauche+)	Met-120 57.4 (gauche-)	Thr-183 -181.0 (trans)	Phe-327 -74.7 (gauche+)

^a Unit: angular degree.

explicit β ARK1 model structure to define the initial position for the automated docking analysis.

Modeling of β ARK1 Conformers. In the β ARK1 model building process, two models were adopted to reflect the flexibility of certain amino acid side chains. Their conformations differ from each other at the following amino acid residues: Ile-197, Leu-271, Ser-334, and Phe-483. The two models were defined as conformer 1 and conformer 2. The χ angle values (reflecting torsion around the C_{α} - C_{β} bond) for each residue of both conformers are summarized in Table 2. Conformer 1 was built during a previous study. This model was based on the crystalline structure of the PKA-PKI5-24 complex registered in the Protein Data Bank as 2CPK and was generated by the homology modeling method in the Protein Design module of the Quanta program.²⁹ To make conformer 2, the χ values of conformer 1 were modified manually to fit with more commonly reported χ angles.³⁰⁻³² This model was defined as conformer 2.

Database Search. The number of candidates that survived each step of the search is presented in Table 3. First, the Converter program³³ was used to generate 3D structures for 265 537 compounds of the original 267 733 compounds contained in the ACD 99.1 database.³⁴ Because the geometry check function of Converter resulted in a failure during the conversion calculation, the structures of 2196 of these compounds could not be generated. Converter also added 53 662 stereoisomers to the new database, making a total of 319 199 entries. Among these compounds, 263 334 had a molecular weight between 200 and 650, the target range. The next step was pharmacophore-based analysis of the database entries by ARCHER. The analysis yielded 134 652 compounds with functional groups that were compatible with the model pharmacophore. Of these 134 652 compounds, 112 314 compounds passed the distance matrix filter analysis. Following these analyses, automated flexible docking of the candidates

Table 3. Overview of the Search Process

search step	candidates remaining	
(1) original ACD	267 733	
(2) successful 3D structure calculation	265 537	
(3) addition of stereoisomers	319 199	
(4) molecular weight filtering (200-650)	263 334	
(5) carrying the pharmacophore's functional group	134 652	
(6) distance matrix fitness	112 314	
(7) successful docking to the β ARK1 model	4387 ^a	11 088 ^b
(8) top scoring 2000 compounds	2000 ^a	2000 ^b
(9) unwanted functional group	3680	614
(10) visual inspection: hydrogen bond formation, stable conformation, and hydrophobic interaction	361	662
(11) uniting multiple entries with the same ID	950	
(12) uniting entries with the similar chemical structure	280	
(13) final selection	11	

^a Docking with the β ARK1 conformer 1 model. ^b Docking with the β ARK1 conformer 2 model.

to the β ARK1 model was conducted by ARCHER. In this step, two conformers of β ARK1 were used, and the calculations were named run 1 and run 2, after their respective conformers. The number of successfully docked compounds from run 1 was 4387 and from run 2 was 11 088. These docking models were evaluated by van der Waals and electrostatic energy terms calculated from the CHARMM function.³⁵ The top-scoring 2000 entries from each ARCHER run were put into one group and used in further analysis. Some compounds were contained in the results of both run 1 and run 2. In addition, this new group of 4000 compounds contained stereoisomers. Therefore, these 4000 entries were not necessarily unique chemical compounds but rather the best candidate docking modes with particular conformations and stereo configurations.

In-Depth Analysis. After the ARCHER search, compounds with unwanted chemical groups were discarded, reducing the total number of candidates from 4000 to 3680. Hydrogen bond formation and hydrophobic interaction were estimated by interatomic distances, and the top-scoring 1000 candidates from each category were selected for further analysis. The conformational stability of each candidate was also estimated by force field energy difference between the complexed conformation and freely minimized conformation, and the top-scoring 1000 candidates from this category were selected for further analysis. Compounds in each of the three categories were visually inspected to eliminate candidates without ideal hydrogen bond geometry, hydrophobic molecular surfaces, or torsion angles; this resulted in 361 candidates with good hydrogen bond geometry, 614 candidates with adequate hydrophobic surface area, and 662 candidates with proper torsion angles, respectively. These candidates were pooled, and redundant entries with the same chemical structure were represented by a single entry, reducing the total to 950. Some of these entries shared the same chemical substructures. For example, there were 111 xanthine derivatives, 155 adenine derivatives, 20 guanidine derivatives, and 121 uracil derivatives. Therefore, compounds with the same substructures were represented

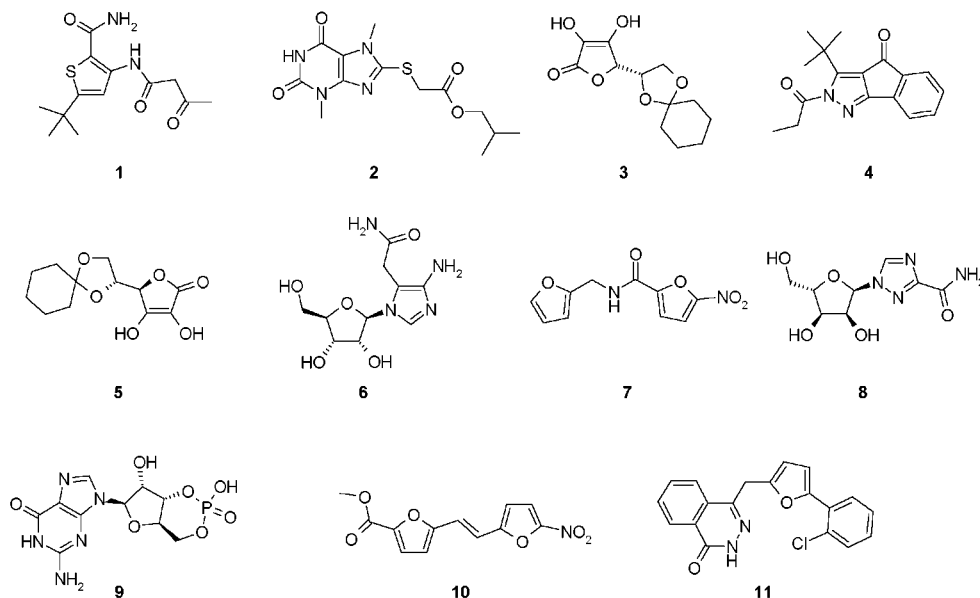


Figure 3. Chemical structures of the 11 candidate compounds selected from the database search for β ARK1 inhibitors.

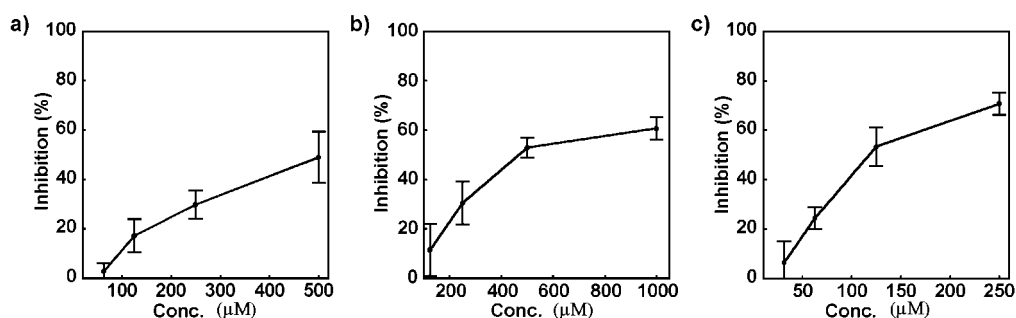


Figure 4. β ARK1 inhibition by (a) compound **4**, (b) compound **6**, and (c) compound **10**. Data are expressed as mean \pm SEM ($n = 6$).

Table 4. Inhibition of β ARK1 Activity Caused by the 11 Selected Compounds and Their Calculated Interaction Energy When Complexed with β ARK1

compd	calcd interaction energy (kcal/mol)	inhibition at 1 mM ^a (percentage of control value)	IC ₅₀ (μ M)
1	-66.289	37.3	
2	-64.148	0.0	
3	-57.274	20.4	
4	-56.988	52.7	563
5	-55.882	0.0	
6	-55.211	60.7	557
7	-54.869	37.7	
8	-53.205	0.0	
9	-50.060	24.2	
10	-49.774	88.4	126
11	-37.878	67.3	

^a Inhibition rate was regarded as zero if the calculated value was below zero.

by the top-scoring compounds. Compounds that were not commercially available were also eliminated. The resulting 280 compounds were inspected visually, and 11 compounds were selected to confirm their ability to inhibit β ARK1 activity in an in vitro assay (Figure 3).

β ARK1 Inhibition Assay. The in vitro β ARK1 inhibitory activities of the 11 selected compounds are listed in Table 4. When tested at 1 mM during the first screening, four compounds inhibited β ARK1 activity by more than 50%. Further assay of these compounds revealed that three compounds, 1-propionyl-8-oxo-9-

tertiarybutyl-indenopirazole (compound **4**), 5-aminoimidazole-4-carboxamide 1- β -D-ribofuranoside (compound **6**), and methyl 5-[2-(5-nitro-2-furyl)vinyl]-2-furoate (compound **10**), have IC₅₀ values in the range from 100 to 600 μ M. None of these compounds have been previously reported to be β ARK1 inhibitors. Additionally, the inhibition curves for these three compounds show concentration dependence (Figure 4), an indication of a real pharmacologic effect. In contrast, none of the randomly selected compounds (Figure 5) inhibited β ARK1 activity by more than 50% at a concentration of 1 mM.

Selectivity Prediction. The graphical drawings for β ARK1 and PKA complexed with different inhibitors show markedly different patterns (Figure 6). The adenine-binding pocket of β ARK1 is large enough to accommodate the candidates without making contact with them. In contrast, in the drawing of the corresponding pocket of PKA, Thr-51 and Glu-127 sterically narrow the pocket or its entrance. The comparative docking study also focused on differences between Leu-271 in conformer 1 of β ARK1 and Met-120 of PKA. Leu-271 of β ARK1 can have two conformations: conformer 1 and conformer 2. In conformer 1, Leu-271 with a gauche(-) χ of 40.1° occupies almost the same space as Met-120 of PKA. In conformer 2, the χ of Leu-271 was modeled in gauche(+) form, -58.2°. Because of this change, the adenine-binding pocket of conformer 2 is larger than the adenine-binding pocket of conformer 1.

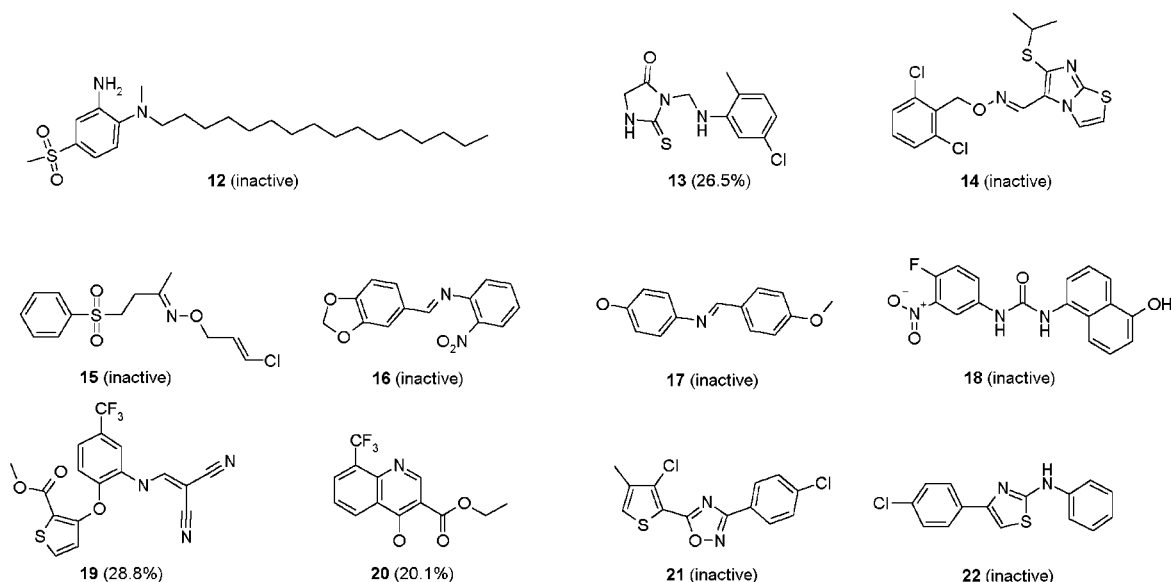


Figure 5. Chemical structures of the 11 compounds selected by a random number generator. The in vitro β ARK1 inhibition activity at 1 mM is given in parentheses.

In model complexes with β ARK1, none of the three molecules came into contact Asp-278, Ile-485, or Arg-199. In contrast to β ARK1, PKA had less space in the binding site to accommodate the drug candidate molecules. The models predicted that compound **4** would come into contact with the side chain of Met-120 and the main chain of Thr-51. Compounds **6** and **10** would come into contact with the side chain of Glu-127. In addition, **10** might lose the CH- π interaction because the Ile-485 of β ARK1 is replaced by Tyr-330 in PKA. Thus, PKA is thought to have an unfavorable interaction with each of the three molecules because of steric differences between β ARK1 and PKA, as shown in Figure 1.

PKA Inhibition Assay. To evaluate the selectivity of these compounds for β ARK1 and not PKA, the in vitro inhibition of PKA activity caused by each of the three selected compounds was assayed (Table 5). These three compounds, **4**, **6**, and **10**, did not inhibit PKA activity even at concentrations of 1 mM. This contrasts with the IC_{50} value of the known PKA inhibitor PKI5-24, which is 0.003 μ M.

Discussion

In the present study, the method applied yielded three inhibitors, **4**, **6**, and **10** (Figure 3), from 11 candidates while none of the randomly selected compounds inhibited β ARK1 at a concentration of 1.0 mM. Computerized visual inspection predicted that none of these compounds would inhibit PKA, and assay results showed that they do not inhibit PKA at a concentration of 1.0 mM. This represents a hit rate for selectivity of 27%. While the best IC_{50} value for this group of compounds is 1.3×10^{-4} M, these compounds are the first low molecular weight inhibitors that can specifically distinguish β ARK1 from PKA. Thus, the results show, at least for the present target protein, that using a search on a database containing 3D structures and a comparative docking study can detect specific protein inhibitors.

At the same time, however, the results suggested room for improvement. For example, the estimated

interaction energy ranking was thought not to parallel the biological activity accurately. There are several possible reasons for this discrepancy. First, the stability estimation of the protein-ligand complex depended on the force field energy function. Generally, such force field energy functions only calculate enthalpy, while entropic terms such as flexibility and hydrophobic interactions might consume a significant portion of the free energy.^{36,37} Better results might be obtained by adopting empirical free energy scoring functions.³⁸⁻⁴⁰ In the present study, however, these functions were not used because the functions cannot be differentiated easily, resulting in larger calculation times. Second, the docking method used in this study did not cover all conformations; therefore, the best docking mode for each molecule might have been overlooked. More precise calculations might yield better results. However, such calculations were not used because the goal of this study was not a precise docking study for each molecule but rather a search to pick up inhibitor candidates by means of a realistic and feasible process of calculation. Third, during the database search, the protein structure that bound the target pharmacophore was assumed to be rigid. This assumption made it difficult to reflect the flexibility of amino acids on the surface of the protein in the final assessment.⁴¹⁻⁴³ Using multiple conformers of the target protein might increase the probability of selecting more compounds, and actually, two models were adopted in the present study. Actually, β ARK1 conformer 1 identified only compound **6** while conformer 2 identified all of the three selective inhibitor leads. Therefore, the use of only conformer 1 would have overlooked compounds **4** and **10**. Thus, an increase in the number of models might lead to more accurate results. Fourth, Because PKA changes its conformation according to bound molecules,⁴⁴ β ARK1 would also change it like PKA does. In the present study, the β ARK1 structure was modeled based on the PKA-PKI binary complex crystal structure in which PKA adopts a "closed" conformation⁴⁵ and the empty ATP binding site is in its native form. Because the virtual screening

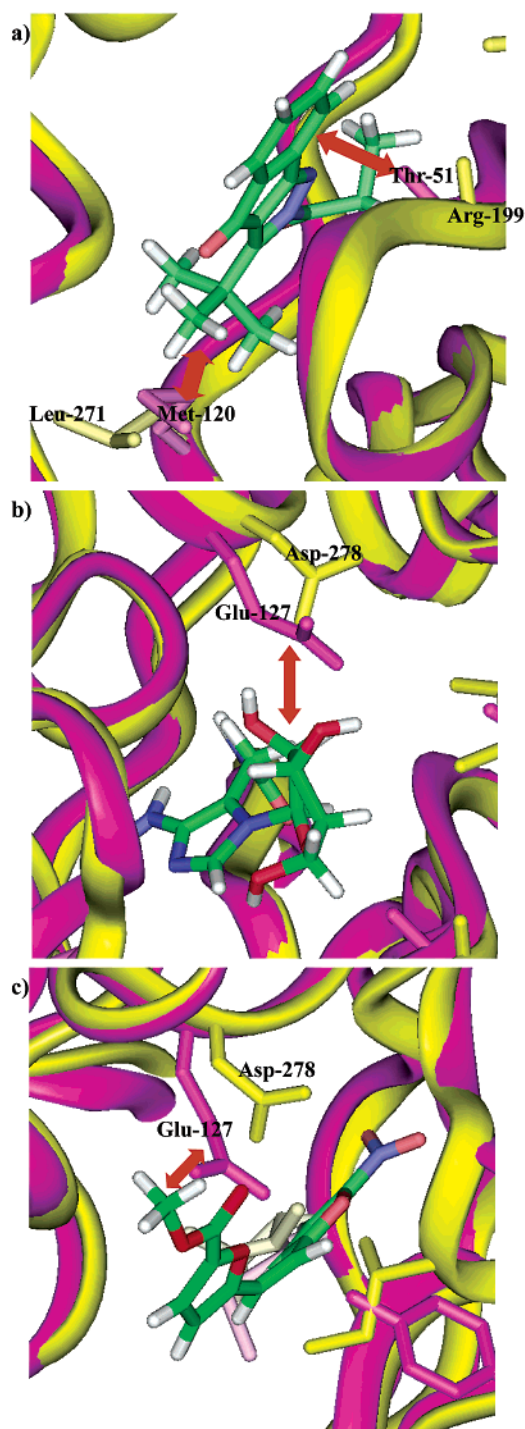


Figure 6. Structural comparison of β ARK1 or PKA in complex with β ARK1 inhibitors. Yellow, β ARK1; pink, PKA. Putative intermolecular repulsion is shown by an orange arrow. (a) Compound **4**, (b) compound **6**, and (c) compound **10**.

using this model identified three selective inhibitor leads, it would reflect a β ARK1 conformation. In addition, β ARK1 models based on other PKA crystal structures would reproduce different conformations of β ARK1, which would diversify the variety of inhibitors resulting from the virtual screening. Fifth, the 3D structure of the target protein was a calculated model because there are no data describing the 3D structure of β ARK1. The accuracy of such models derived from homology modeling depends on how similar the modeled protein is to the reference protein. In the present study, the amino

Table 5. PKA Inhibitory Activity of Compounds **4**, **6**, and **10**

compd	inhibition (percentage of control value)
4	-2.2% at 1 mM
6	-6.5% at 1 mM
10	1.8% at 1 mM
PKI 5-24 ^a	IC ₅₀ = 0.003 μ M

^a Positive control compound.

acid identity of the modeled region between β ARK1 and PKA was 34.4%. With such a relatively low identity, the modeled structure might have inaccurate side chain orientations or main chain folding,⁴⁶ which would deleteriously affect the overall rational drug design process. Finally, it is also possible that other regions not modeled play a significant role in ligand binding.

In the step following the 3D database search, the binding specificity of the inhibitor for β ARK1 and PKA was examined visually based on the structures of the putative protein–compound complexes. There were marked steric differences between the ATP binding site of the β ARK1 model structure and PKA crystalline structure. These differences were important to the present study since intermolecular contact could be predicted through a graphical representation technique such as molecular surface drawing. The comparison between β ARK1 and PKA through superposition presupposes that all inhibitors of these proteins that act at this site have a common binding mechanism. This is a likely assumption since there are several examples in which the binding mechanism of a ligand is conserved between proteins with similar structures and functions.^{47,48} However, the visual comparison process was qualitative and not quantitative. This process should be replaced with an objective method as soon as a scoring function with enough accuracy is developed.

The activity of the inhibitors obtained in the present study is not as potent as the activity usually observed for medicines. Their physical and chemical properties were also not favorable for clinical use because of moieties such as nitro groups or sugar substructures. Thus, the next step is to optimize the chemical structures of the compounds to achieve higher affinity for β ARK1 and better bioavailability. Because the putative 3D structural information is already available for complexes between β ARK1 and the inhibitors, the modification process to increase affinity and selectivity will likely be rapid. The process of molecular design to obtain favorable properties will probably involve remodeling of the tertiary butyl group of compound **4**, replacement of the sugar scaffold of compound **6**, and substitution of the nitro group of compound **10** by other chemical groups with the similar geometry and interaction properties. As these compounds are modified, the models of the complexes will provide the geometric information necessary to make appropriate changes.

Conclusions

In conclusion, specific β ARK1 inhibitor candidates were obtained from a search of a database containing 3D drug candidate structures, followed by a comparative docking study. The inhibitor candidates resulting from this screening inhibited β ARK1 without affecting PKA activity. Indeed, these are the first small size selective inhibitor leads of β ARK1 ever reported. The results of

the present study demonstrate the effectiveness of a combined strategy of 3D database search and a consecutive comparative docking study for identifying selective inhibitor leads for β ARK1 and indicate its general usefulness for other systems.

Experimental Section

Materials. PKA and a colorimetric PKA assay kit were purchased from Pierce (Rockford, IL). The PKA inhibitor PKI5-24 was purchased from Biomol Research Laboratories, Inc. (Plymouth Meeting, PA). The structures of the compounds assayed are shown in Figures 3 and 5. Compounds **1**, **7**, **13**, **19**, **21**, and **22** were purchased from Maybridge (Cornwall, U.K.); compounds **2**, **12**, **16**, and **18** were purchased from SALOR (Sigma Aldrich Library of Rare Chemicals; Aldrich Chemical Co. Inc.; Milwaukee, WI); compound **3** was purchased from Aldrich; compound **4** was purchased from Menai (N. Wales, U.K.); compound **5** was purchased from Fluka (Milwaukee, WI); compounds **6**, **8**, and **9** were purchased from Sigma (St. Louis, MO); compounds **10**, **11**, **14**, **15**, and **20** were purchased from Bionet (Cornwall, U.K.); and compound **17** was purchased from TCI (Tokyo, Japan).

Hardware and Software. Four personal computers (each having two Pentium II 450 MHz processors) running the Red Hat Linux 5.2 operating system were used to develop and run the ARCHER program. A Silicon Graphics Octane (R10000 195 MHz processor) computer running the IRIX 6.5.3 operating system was used to draw and manipulate molecular structures. An IBM RS/6000 running the AIX 4.1.4 operating system was used as the ISIS/Host 2.2.1h server, and a PC (Pentium 120 MHz processor) running Windows 95 was used as the ISIS/Base 2.2.1h client. The ISIS/Base⁴⁹ program was used as the database filtering software. The Quanta 97 program was used to calculate Connolly surfaces to visualize molecules and to draw molecular structures. The Insight II 98.0 program⁵⁰ was used to fill hydrogen and to run the CHARMM program, and the Converter program, which is a module of the Insight II software suite, was used to generate the 3D structures. The Cerius 2 program⁵¹ was used to group molecules. To filter and search compounds in the chemical database, an in-house program named ARCHER (Automatic and Rational Complementary Hit-compound Explorer) was developed.

Pharmacophore Generation. To streamline the initial docking calculation between β ARK1 and candidate compounds, the complexity of the putative target site on β ARK1 was simplified into a pharmacophore model consisting of elements such as HBDs, HBAs, and ARCs, based on the following observations. In the adenine-binding pocket of PKA, the amide oxygen of Glu-121 and the amide nitrogen of Val-123 form strong and specific hydrogen bonds with ATP.²⁴ Therefore, the corresponding atoms in β ARK1, which were the amide oxygen of Asp-272 and the amide nitrogen of Met-274, were used to define pharmacophore points in β ARK1 for hydrogen bonds similar to those observed in a PKA–ligand complex. For the amide oxygen of Asp-272, an HBD was placed 2.8 Å, a standard hydrogen bond length, from the oxygen in the sp² plane with the lone electron pair pointing in the direction of the oxygen. For the amide nitrogen of Met-274, an HBA was placed 2.8 Å, a standard hydrogen bond length, from the nitrogen in the sp² plane with the hydrogen bond pointing in the direction of the nitrogen. Then, to remove short contacts, the position of each HBD and HBA was optimized while the β ARK1 position remained fixed through a CHARMM force field calculation. In the calculation, the HBD was assumed to have the characteristics of an sp³ nitrogen atom, and the HBA was assumed to have the characteristics of an sp³ oxygen atom. Additionally, in the adenine-binding pocket of PKA and other kinases, an aromatic ring system is recognized specifically.⁵² Assuming the same recognition mechanism operates in β ARK1, an ARC was placed by referring to aromatic ring center coordinates of other known kinase–ligand complex structures in the PDB such as 1BKX, 1STC, 1YDS, and 1YDT. Then, to remove unfavorable contacts between the compounds and the

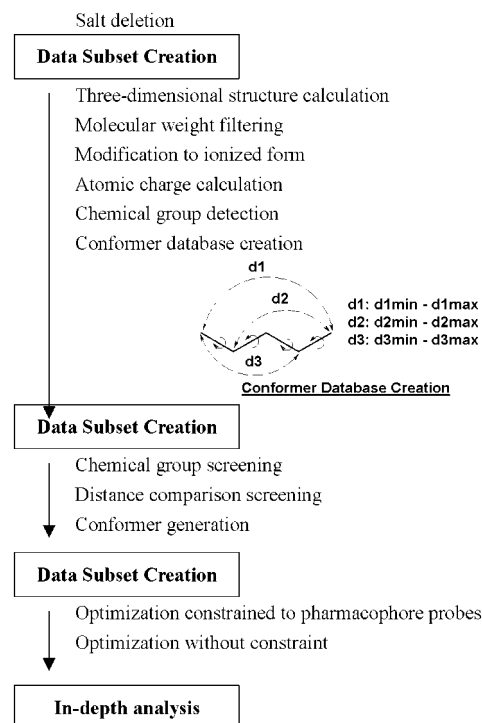


Figure 7. Outline of ARCHER procedure.

enzymes, the position of the ARC was optimized while the β ARK1 position remained fixed through a CHARMM force field calculation. In the calculation, the ARC was assumed to have the characteristics of an sp³ carbon atom.

Database Search. The pharmacophore search and protein-docking analysis algorithms of the ARCHER program were derived from those of ISIS/Base¹² and other previously reported programs^{13–15} with some modifications. Figure 7 shows the outline of the ARCHER calculations. The ARCHER procedure consists of three steps. The first step is initial candidate selection by filtering and structure modifications that include 3D structure generation; the second step is a pharmacophore-based conformationally flexible database search; and the third step is a docking study using the target proteins. Because of licensing restrictions, all files containing molecular information were saved only temporarily and were used only as a means of passing information between programs.

In the first step, a data subset was created by the following filtering process. The Available Chemical Directory (ACD) 99.1 was used as the database of commercially available compounds.⁵³ Unique molecules were isolated by taking only the largest molecule for any particular formula entry while other molecules in the same entry were regarded as salt molecules and were removed. After these redundant salts were removed, the Insight II/Converter software suite was used to fill in all possible hydrogen atoms and convert the formula into a 3D structure. For compounds with several racemic centers, at most four arbitrary stereoisomers were generated and each resulting structure was treated as a new database entry. Molecules for which a 3D structure could not be calculated by Converter were discarded. A molecular weight range from 200 to 650 was defined as the preferred range for drug molecules and was also used as a cardinal condition of a filtering process. Acidic functional groups such as carboxylic acid, sulfonic acid, and phosphoric acid were deprotonated, and basic groups such as amines, amidines, and guanidines were protonated. Gasteiger's atomic charges⁵⁴ were also calculated. Finally, possible conformations were generated systematically but temporarily by rotation of freely rotatable bonds. During conformation generation, all pairs of interatomic distances between HBD, HBA, and ARC were recorded. Then, a database of distance ranges for every possible combination of HBD, HBA, and ARC was created.

In the second step, a pharmacophore search was performed on this initial candidate group. The search consisted of the following procedures. First, the pharmacophore elements were used to select qualified compounds that possess the needed number and kind of functional groups for the current pharmacophore, using the key screening algorithm in the ISIS/Base program¹² and the formula screen algorithm in the ChemDBS-3D program.¹³ At this stage, molecules with at least two of the pharmacophore elements were selected. Second, the upper and lower distance range database of resulting candidates was screened with the distance matrix of the model pharmacophore, revealing whether a candidate compound was comparable with the pharmacophore. This method was inspired from the minimum residual algorithm used in the FLOG program.¹⁴ Third, conformers were generated systematically for each candidate and superimposed onto the pharmacophore model by rigid body rotation and translation. The goodness-of-fit of each conformer to the model pharmacophore was assessed by root mean square distance calculations between the pharmacophore elements and the corresponding points of the conformer.

In the third step of the filtering process, force field calculations were applied to minimize the energy of each conformer generated to fit the β ARK1 pharmacophore model. ARCHER scored each compound using eq 1 and 2

$$E = \sum_{i,j>i} \frac{q_i q_j}{4\pi\epsilon_0 r_{ij}^2} + \sum_{i,j>i} \epsilon_{ij}^* \left[\left(\frac{r_{ij}^*}{r_{ij}} \right)^{12} - 2 \left(\frac{r_{ij}^*}{r_{ij}} \right)^6 \right] \quad (1)$$

$$\epsilon_{ij}^* = (\epsilon_i^* \epsilon_j^*)^{1/2}, r_{ij}^* = r_i^* + r_j^* \quad (2)$$

where q is the atomic charge, r is the interatomic distance, ϵ_0 is the dielectric constant, and ϵ is the van der Waals constant. An asterisk means a standard value of the force field. The force field parameters and functions were taken from the CHARMM program. The grid energy calculation method was used to calculate intermolecular interaction energy, and β ARK1 was treated as a rigid protein. To reflect the flexibility of certain amino acid side chains, two conformational models of β ARK1 were constructed. Search calculations using each model were named run 1 and run 2, respectively. Energy minimization of each candidate- β ARK1 complex consisted of two phases. The first phase was energy minimization of the candidate compound constrained by the superimposed pharmacophore, and the second phase was nonconstrained minimization. Ligand translation was limited within the extents of the grid, similar to the function used in the AutoDock program.¹⁵ After these minimization processes, the 2000 top-scoring candidates from run 1 and run 2 proceeded to further analysis. These 4000 candidates contained the entries with the same chemical structure but different chiral center configurations and different docking orientations.

In-Depth Analysis. Candidates resulting from the ARCHER search were analyzed further. First, candidates with unwanted functional groups such as acid halides, acid anhydrides, aldehydes, azides, isocyanates, thiocyanates, epoxides, and thiols were eliminated. Then, each candidate was evaluated with the following three measurements: (i) the distances of hydrogen bonds formed by the amide oxygen of Asp-272 or formed by the amide nitrogen of Met-274, (ii) the number of aliphatic and aromatic carbons within 4.0 Å from side chains of Ile-197, Val-205, Ala-218, Met-274, Leu-324, and Phe-483, and (iii) the force field energy difference between the complexed conformation and the minimized conformation calculated from the complexed conformation. The top-scoring 1000 candidates from each category were then inspected to eliminate candidates with unrealistic hydrogen bond geometry, discontinuous hydrophobic molecular surfaces, or unstable torsional configuration, respectively. The candidates in each category were united as a group. Then, entries selected from both run 1 and run 2, redundant stereoisomers, and entries that are not commercially available were eliminated. Compounds with similar chemical structures were also eliminated

after clustering by the Cerius2 program and substructure search by the ISIS/Base program. The resulting candidates were visually inspected on the computer display to make the final selection.

Negative Control. As a negative control group, 11 compounds were selected randomly from the 263 334 entries, which was the entry list just before the pharmacophore search. A random selection script using the rand function written in Perl 5.005 was run on Red Hat Linux 6.2 to select the compounds.

β ARK1 Inhibition Assay. The inhibition of β ARK1 activity caused by each test compound was assayed using a previously reported method, with some modifications.⁵⁵ A reaction mixture containing 10 mM [γ -³²P]ATP (18.5 kBq), the intracellular carboxyl terminal domain of the human β 2-adrenergic receptor (6.7 μ M), and either one concentration (range: 2.0–6.0 μ g/mL) of β ARK1 or a blank solution (as a control) were incubated in reaction buffer (20 mM Tris-HCl, pH 7.5; 5 mM MgCl₂; 2 mM EDTA; 0.5 mM EGTA) at 30 °C for 60 min. After the phosphorylation reactions were stopped by immersing the test tubes in ice water, electrophoresis was used to separate the components of the reaction mixture, and an Imaging Plate (Fujifilm, Japan) was used to record the distribution pattern of radioactivity on the electrophoresis gel. A BAS 2000 bioimage analyzer (Fujifilm, Japan) was used to read the Imaging Plate. During the initial screening, the concentration of each compound was 1 mM and only one reaction was conducted per candidate compound. The secondary screening was conducted with six replicates of four different concentrations for each candidate compound. The percentage inhibition rate was calculated by eq 3

$$\text{inhibition} = \frac{R_{\text{control}} - R_{\text{sample}}}{R_{\text{control}} - R_{\text{background}}} \times 100 \quad (3)$$

where R_{sample} , R_{control} , and $R_{\text{background}}$ are the radioactivity in the assayed sample, positive control, and background samples, respectively. The percentage inhibition rate was recorded as zero if the calculated value was below zero.

Selectivity Prediction. After finding potential β ARK1 inhibitors, their ability to inhibit PKA was estimated by superimposing the crystalline structure of PKA, registered in the Protein Data Bank as 2CPK, onto the β ARK1-candidate complex model in the same coordinate system. Then, a structural comparison, which has been successfully applied to estimate ligand selectivity,^{56,57} was applied to β ARK1 and PKA. The major steric difference between Asp-278 of β ARK1 and Glu-127 of PKA was visualized by means of a Connolly surface image (probe radius: 1.4 Å). Because force field energy depends on the number of atoms included in the system, absolute force field energies could not be compared between PKA and β ARK1. Therefore, a comparative docking study and visual examination were used to perform qualitative analysis instead of using a quantitative assessment such as energy calculation. Using interatomic distance information and atomic charge information, candidates with short contact with the enzyme except for hydrogen-bonding atoms or with electrostatic pattern mismatches were judged to be unfavorable for PKA binding.

PKA Inhibition Assay. The inhibition of PKA activity caused by each test compound was assayed using a colorimetric PKA assay kit. PKI5-24 was used as the positive control inhibitor for PKA. The assay was performed according to the directions supplied with the kit. A test compound, PKA (final activity: 87 units/ μ L), and the dye-linked PKA substrate (kemptide, sequence: LRRASLG) were incubated in reaction buffer (2 mM ATP, 10 mM MgCl₂, 0.002% Triton X-100, 20 mM Tris(hydroxymethyl)amino methane, 100 mM cAMP, 5% DMSO, pH 7.4) for 30 min at 30 °C. After the reaction was terminated, the phosphorylated dye-linked substrate was selectively trapped by an affinity membrane, was isolated by means of the elution buffer supplied with the kit, and was quantified by OD measurement at 570 nm. Optical density was

measured by means of a Spectra Max 340 spectrophotometer (Molecular Devices Corp., CA).

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Supporting Information Available: Coordinates for each candidate (compound 1–11) and adenine-binding pocket of β ARK1 conformer 1 and conformer 2. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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